

## TOWARD HIGHLY STABLE [<sup>212/203</sup>Pb]Pb<sup>2+</sup> COMPLEXES: RIGIDIFIED HEXADENTATE LIGAND DESIGN

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**Aim/Introduction:** The *theranostic* approach exploits the same drug delivery system with either diagnostic or therapeutic radionuclide, allowing an accurate prediction of the therapeutic biodistribution. Lead-203 (<sup>203</sup>Pb, *t*<sub>1/2</sub> 51.9 h, γ, 80.94 %) and lead-212 (<sup>212</sup>Pb, *t*<sub>1/2</sub> = 10.6 h, β<sup>-</sup>, 100% followed by a decay chain including α-emission) are one example of a *true theranostic pair* [1]. This study reports two hexadentate acyclic chelators based on a rigid *trans*-diaminocyclohexane (DACH) scaffold [Figure 1] to improve both thermodynamic stability and kinetic inertness of the Pb<sup>2+</sup> complexes.

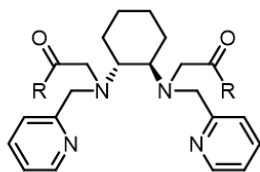


Figure 1. Structure of the investigated DACH-based ligands.  
R = OH (*H<sub>2</sub>bpcd*); NH<sub>2</sub> (*NH<sub>2</sub>bpcd*).

**Materials and Methods:** Thermodynamic and kinetic studies were performed combining potentiometry, UV-Vis and <sup>1</sup>H NMR spectroscopy. <sup>203</sup>Pb was produced via the <sup>203</sup>Tl (p,n)<sup>203</sup>Pb reaction at TRIUMF's TR13 cyclotron. Concentration-, time-, temperature-dependent radiolabelling with <sup>203</sup>Pb were conducted at pH 6 (0.1 M NH<sub>4</sub>OAc). The radiochemical yields (RCYs) were determined by radio-TLC. The *in vitro* stability of the preformed <sup>203</sup>Pb complex was assessed in human serum over 48 h by radio-TLC.

**Results:** *H<sub>2</sub>bpcd* and *NH<sub>2</sub>bpcd* combine pyridine and carboxylate/carboxamide donor groups creating an N<sub>4</sub>O<sub>2</sub> coordination environment. NMR complexation studies with <sup>nat</sup>Pb<sup>2+</sup> show that both ligands have good affinity for Pb<sup>2+</sup>, revealing rapid complex formation and high stability in acidic conditions (pH ~1). However, the two complexes differ in the isomerization profiles: *H<sub>2</sub>bpcd* forms one dominant species in solution, while *NH<sub>2</sub>bpcd* yields two isomers. These results are further confirmed by DFT calculations. Thermodynamic studies reveal a significantly higher thermodynamic stability of *H<sub>2</sub>bpcd* compared to *NH<sub>2</sub>bpcd*, with the former establishing a new benchmark among acyclic ligands for Pb<sup>2+</sup> (pPb<sup>2+</sup> 15.9 vs 9.7 for DTPAm and 14.8 for *H<sub>2</sub>ampa*). [<sup>203</sup>Pb]Pb<sup>2+</sup> radiolabeling experiments show that only *H<sub>2</sub>bpcd* can incorporate the radiometal (RCY% > 95%, [*H<sub>2</sub>bpcd*] = 10<sup>-5</sup> M, pH 6, 15 min, RT/80°C) while *NH<sub>2</sub>bpcd* shows poor radiolabeling efficiency.

**Conclusions:** This study pinpoints *H<sub>2</sub>bpcd* as a promising scaffold capable of forming highly stable <sup>nat</sup>Pb<sup>2+</sup> complexes and efficiently binding [<sup>203</sup>Pb]Pb<sup>2+</sup>. These findings provide strong support for further (pre)clinical translation of <sup>212/203</sup>Pb-labelled radiotracers.

[1] A. Ingham, T. I. Kostelnik, B. L. McNeil, N. Choudhary, B. O. Patrick, M. de Guadalupe Jaraquemada-Peláez and C. Orvig, *Dalton Trans.*, **2021**, 50, 11579